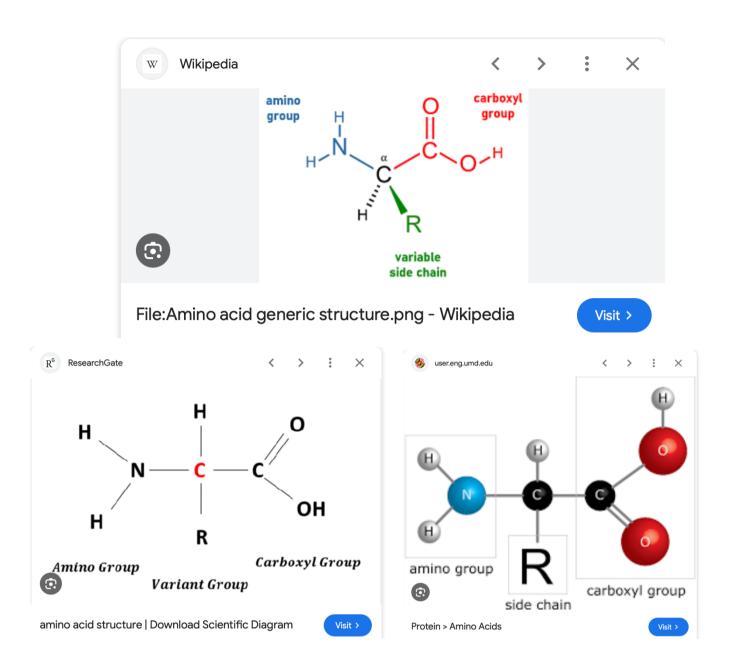
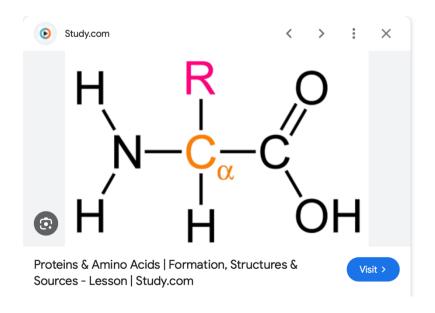
Amines -> Relatively strong bases
and relatively strong
nucleophiles

 $R-NH_2+H\Phi \Longrightarrow R-NH_3$ present at neutral pH =10 pH (pH=7)

Amines are protonated and positivelycharged at neutral pH => Very important in biochemistry!



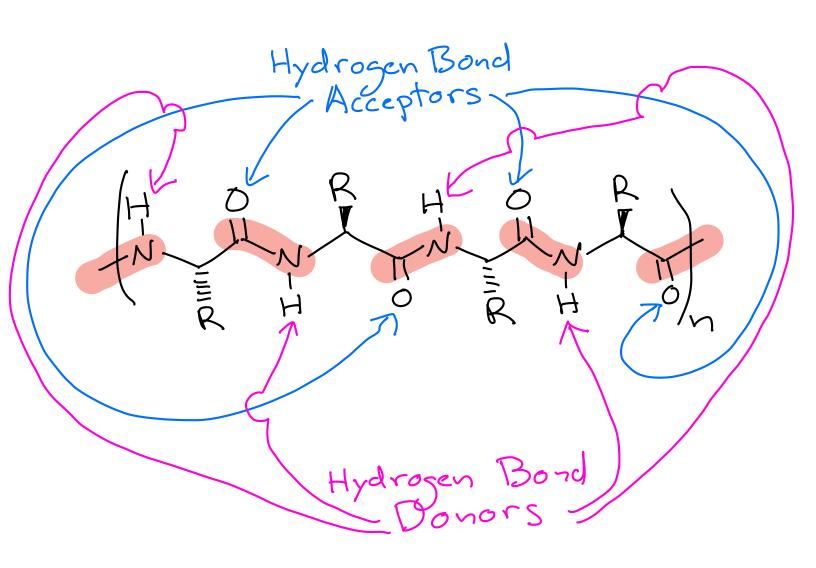


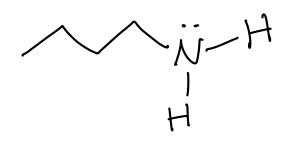


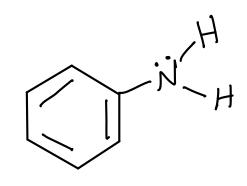
What does this means for amides:

1) The amide group can make strong hydrogen bonds

2) The C-N bond does not rotate at room temperature







Golden rule:

N' electrons

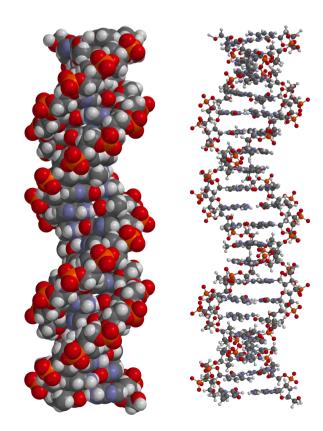
are more

stable when

delocalized

This is critical to DNA and RNA structure: DNA bases are aromatic and the -NH2 groups on the bases are

G·C base pair



Our sense of smell is highly sensitive to certain molecules that are the result of decomposition of mammal and fish flesh among other things. Not only can we detect very small amounts of these "signal" molecules, we are hard wired to be highly nouseated when we smell them sevolutionary protection to keep us from eating what might look ok, yet would make us sick.

H₂Ni / Ni H₂

H₂S

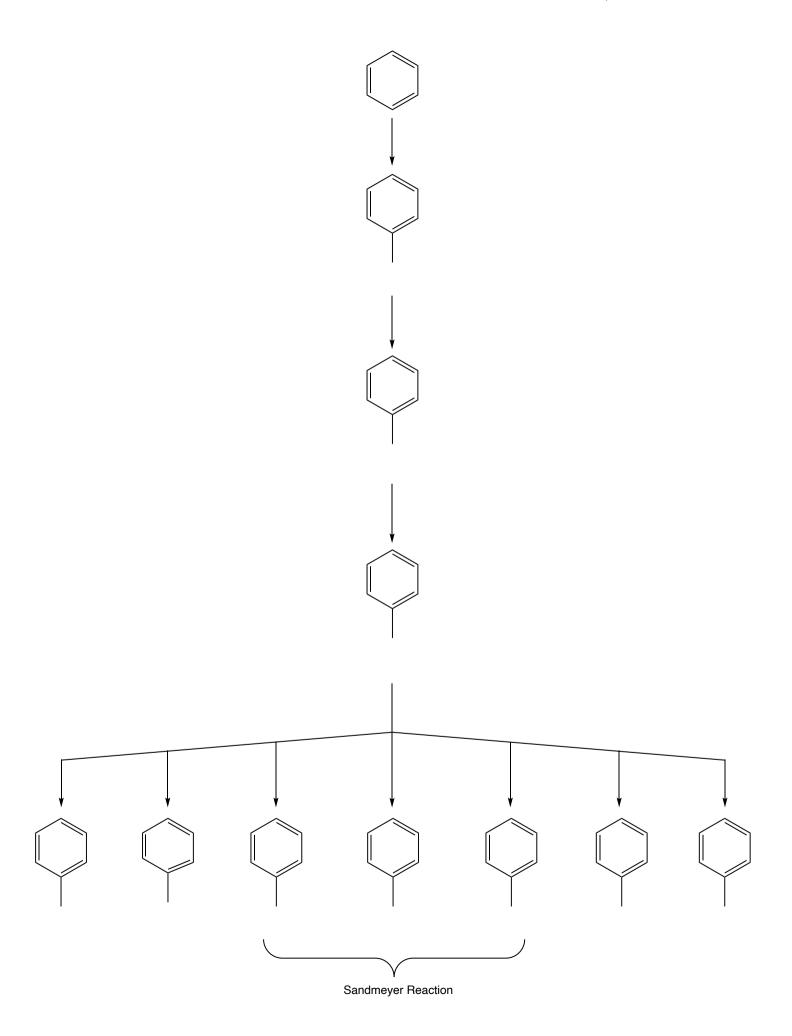
OH

	Strongly activating	−ÑH₂	—ÑHR	$-\ddot{N}R_2$	— <u>ё</u> н	−ÿR
Ortho-Para Directing	Moderately activating	O — NHCR	-NHCAr	−öcr	−öcAr	GOOD
Ortho-Pa	Weakly activating	—R	$\overline{}$		OF I	tho, para directing activating
	Weakly deactivating	− <u>;</u> :	-çı:	<u>ë</u> r:	— <u>;;</u> :	ortho, para directing UGLY
Meta Directing	Moderately deactivating	—СН	O - - - -	—coн	-cor	$-\text{CNH}_2$ $-\text{SOH}$ $-\text{C}\equiv\text{N}$
_						Ü

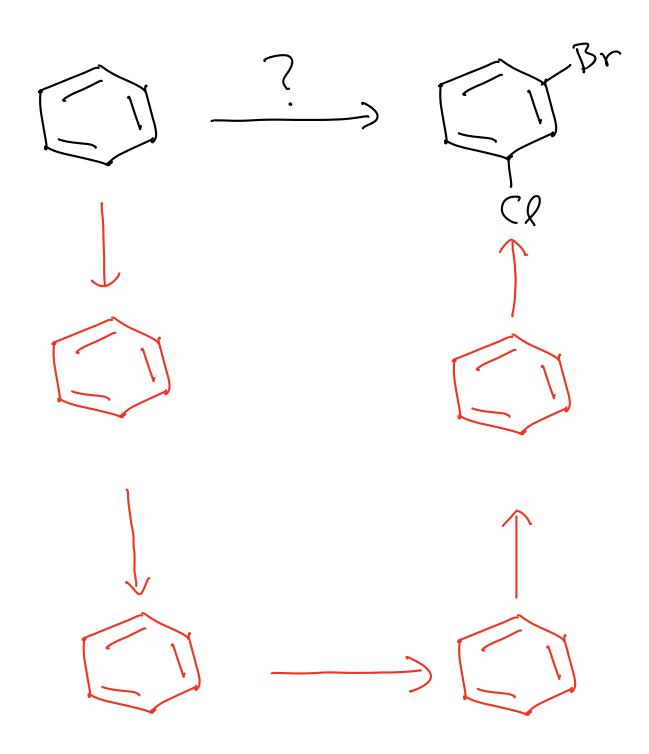
The order in which you add groups matters!

How do we carry out the following synthesis?

Preparation of Diazoniums, The "Mr. Bill" Reaction



How do we carry out the following synthesis?



$Nucleophilic\ Aromatic\ Substitution$

What you need to know about electrophilic aromatic substitution reactions:

1) Friedel-Crafts alkylations and acylations

AP(l3

However -> Some reactions with harsh conditions will work:

H2SOY
(excess)

Toluene ~

Solvent used in model glue

Carbohydrates

Monosaccharides -> 5 or 6 carbons and are aldehydes and ketones

Ex. Glycose is an

Stereochemistry

$$\begin{array}{ccc} CHO \\ = \\ - \\ CH_{2}OH \end{array}$$

(R)-D-Glyceraldehyde

Called Fischer projection

Can I have a hug please?

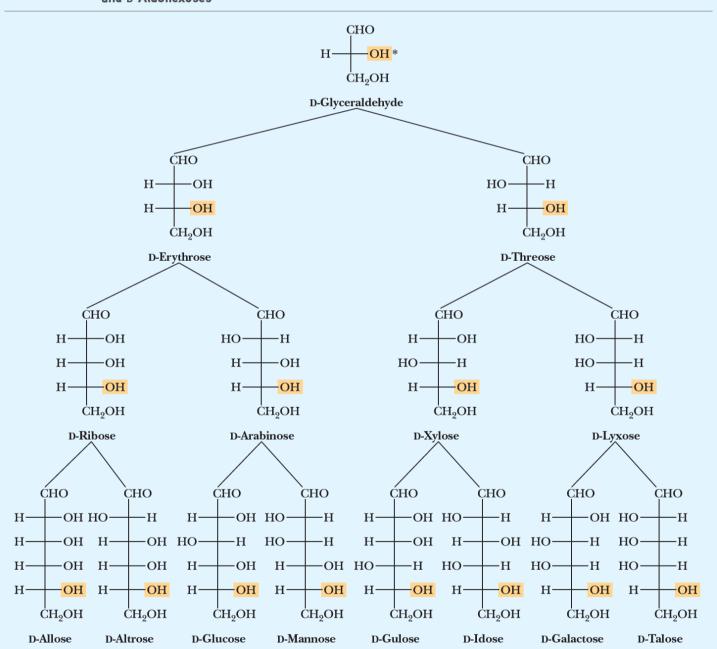


Stereochemistry defines the
different carbohydrates

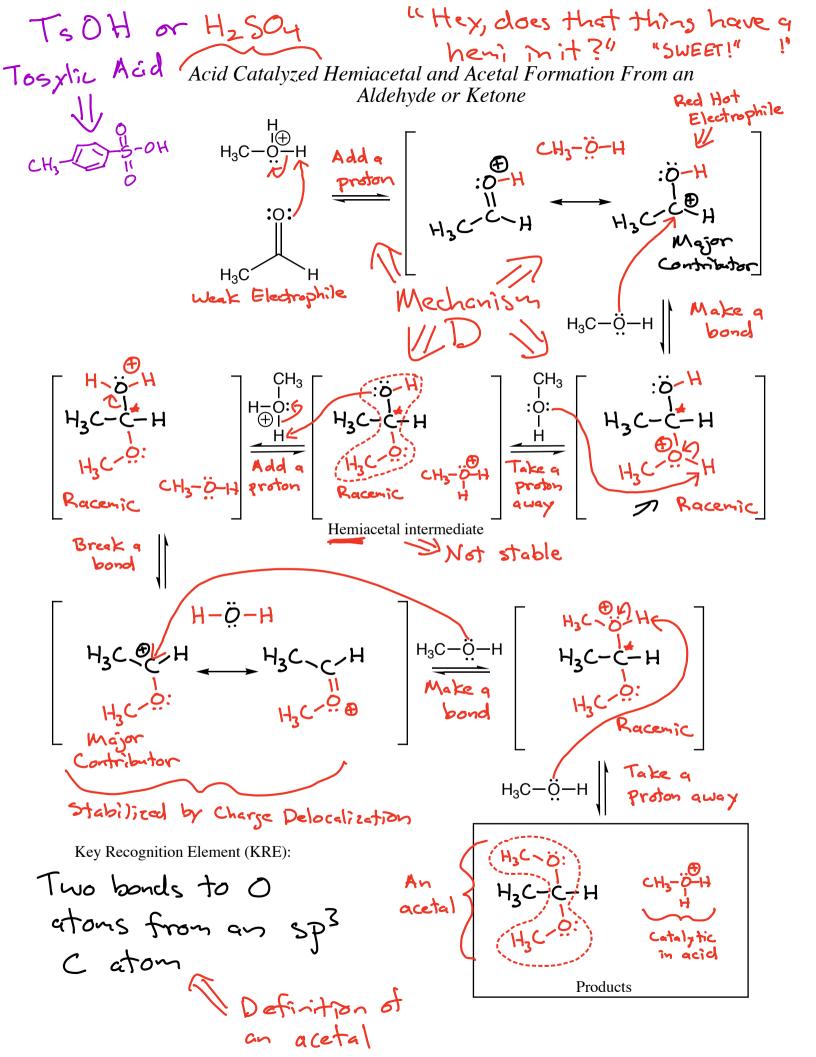
Due to the way carbohydrates
are synthesized in cells ->
the common carbohydrates
all have the same stereochemistry
as D-glyceraldehyde at the
carbon farthest from the
carbonyl (aldehyde or ketore)

That is why they are called "D" carbohydrates

Table 25.1 Configurational Relationships Among the Isometric D-Aldotetroses, D-Aldopentoses, and D-Aldohexoses

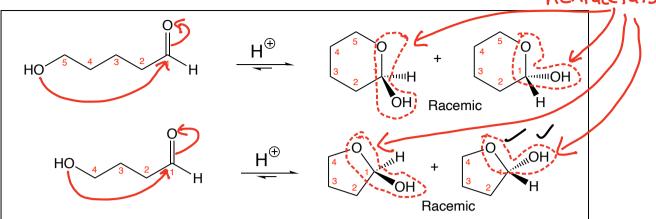






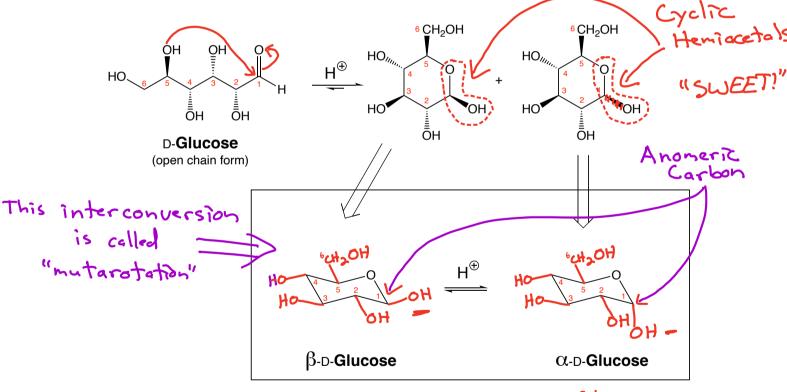
Cyclic Hemiacetals and Carbohydrates

Cyclic hemiacetals



The cyclic form of hemiacetals are stable-"SWEET!"

The chelate effect



Biochemists call these two forms B-D-Glucopyranose
means "6-mebere
ring"

More stable ->
every group is
equatorial!

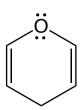
a-D-Glucopyranose

Less stable ->
one -OH is
exial

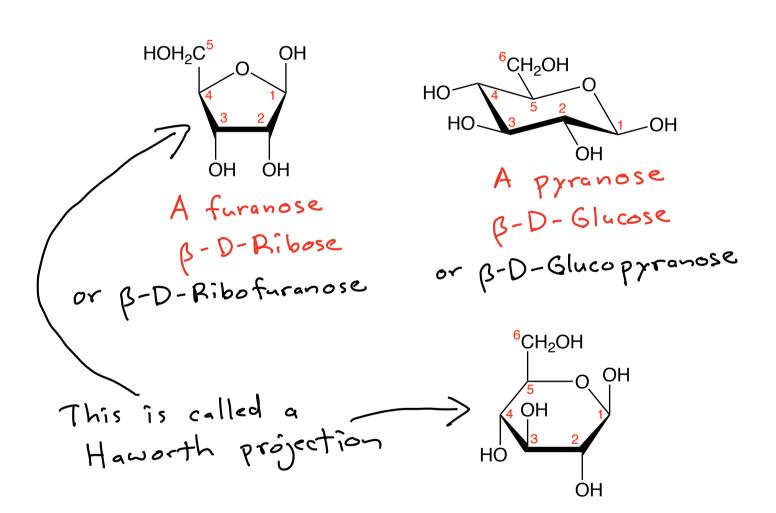
For Glucose:
"X is axial!"



Furan

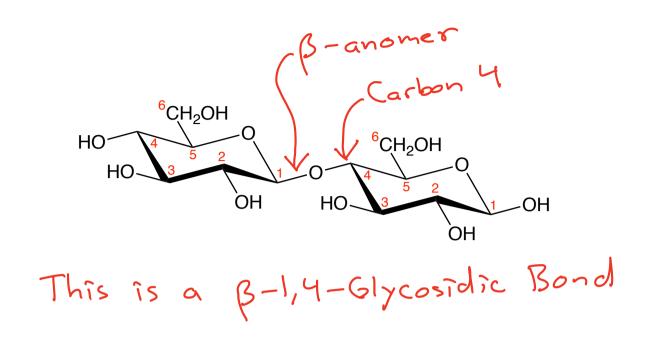


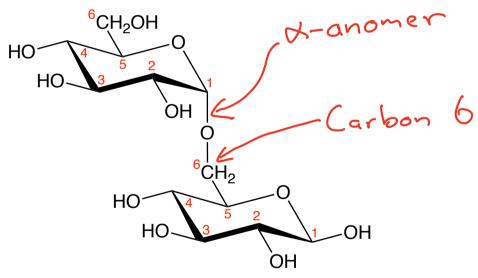
Pyran



N-Acetyl-D-Glucosamine (GlcNAC)

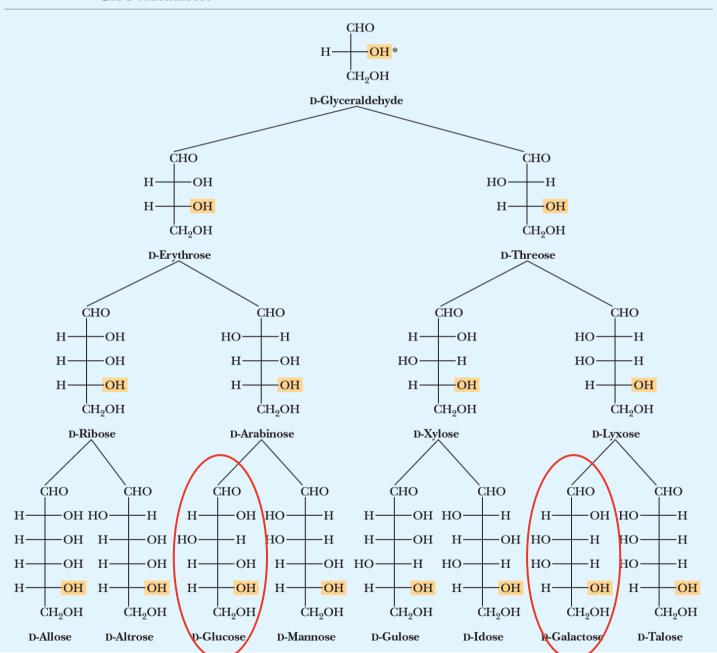
Acetal product

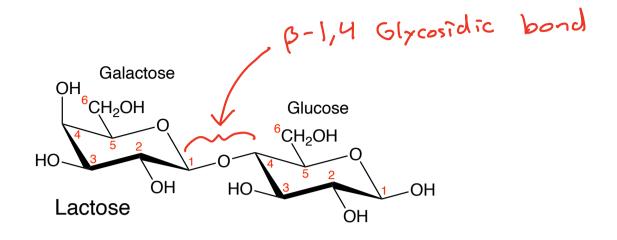


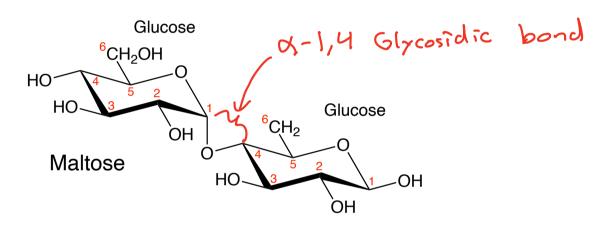


This is an a-1,6-Glycosidic Bond

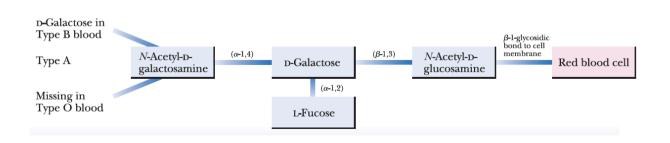
Table 25.1 Configurational Relationships Among the Isometric D-Aldotetroses, D-Aldopentoses, and D-Aldohexoses







We can link more carbohydrates together, always at Carbon 1, with dor Blinkages at carbons 2,3,4 or 6

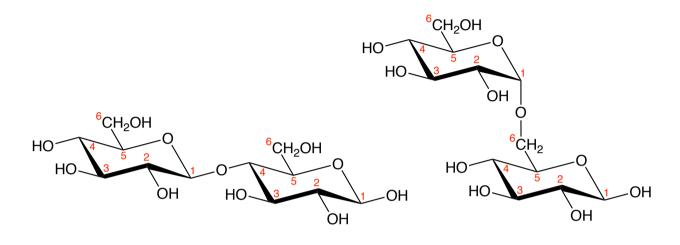


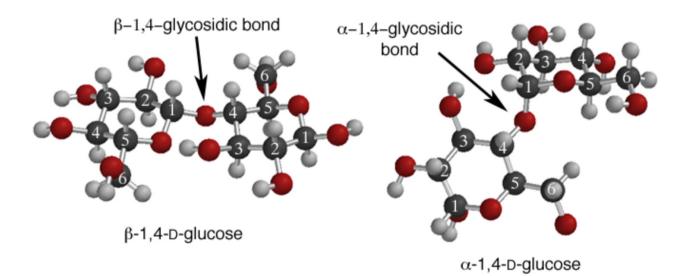


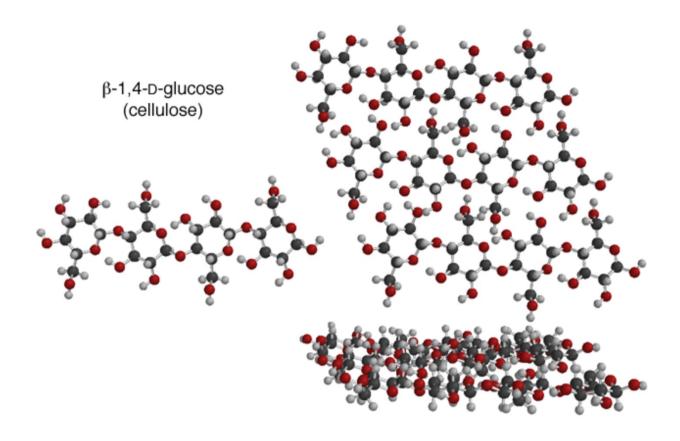
V5.

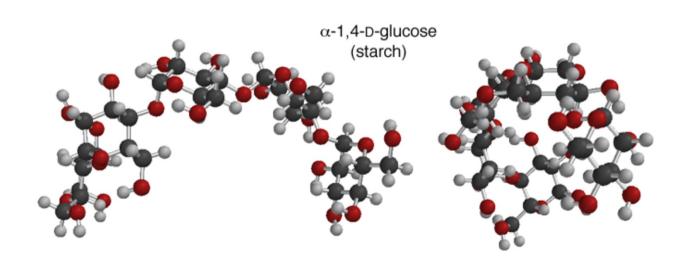


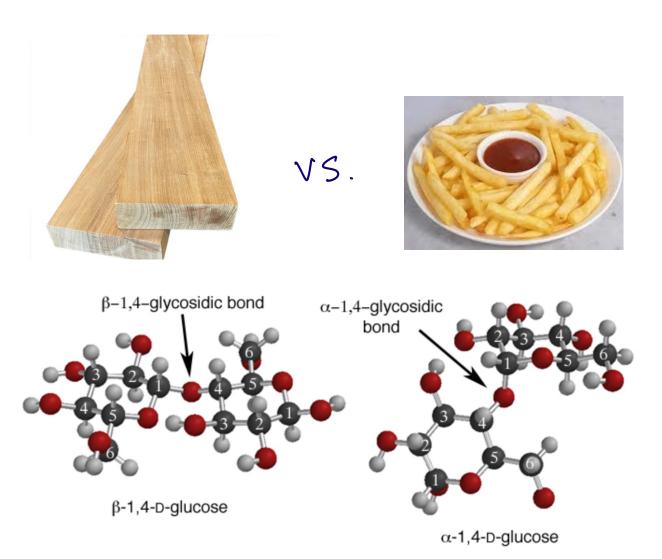
What is the difference — They are both polymers of D-Glycose

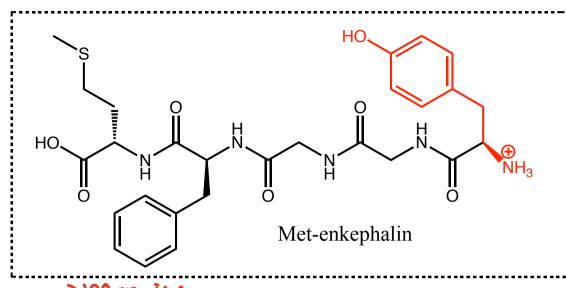


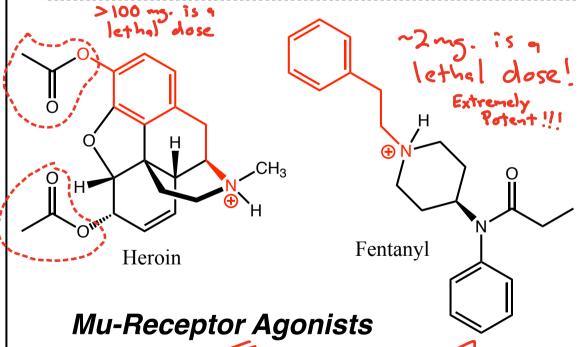




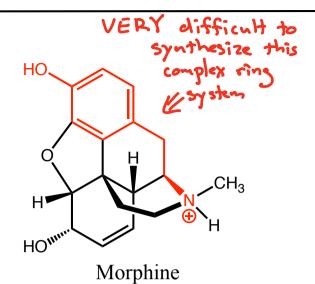




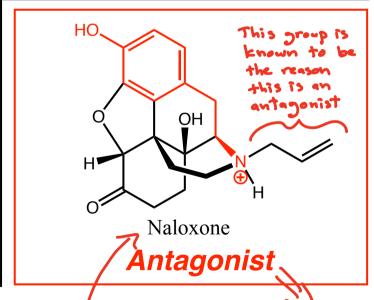




Means it binds (Too easy! SVERY easy and activetes Too easy! I to make!



Isolated from poppies grown in the Middle East



Synthesized from natural morphine analog in a couple of easy steps

Means it binds to receptor but does activate it

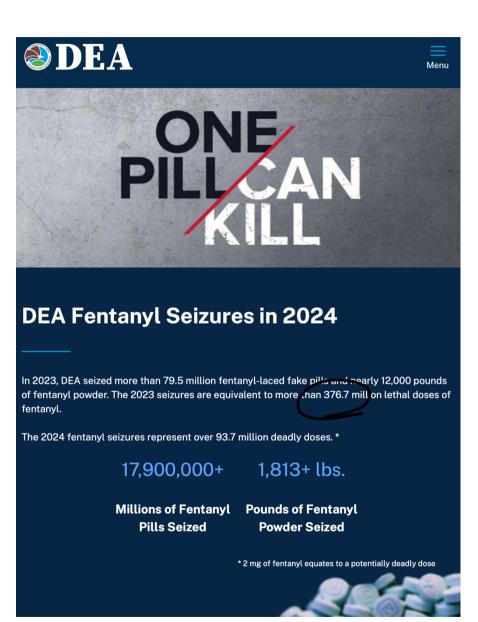
HEALTH

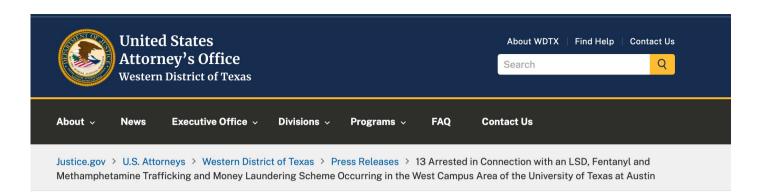
Why fentanyl is deadlier than heroin, in a single photo

By Allison Bond Sept. 29, 2016

Reprints







PRESS RELEASE

13 Arrested in Connection with an LSD, Fentanyl and Methamphetamine Trafficking and Money Laundering Scheme Occurring in the West Campus Area of the University of Texas at Austin





News [Sports] Hookem.com Austin360 Opinion Advertise Obituaries eNewspaper Legals

Jake Ehlinger's family releases statement saying Texas player died of accidental overdose



Brian Davis Hookem

Published 4:56 p.m. CT Oct. 21, 2021 | Updated 3:54 p.m. CT Oct. 22, 2021











Remembering Texas linebacker Jake Ehlinger

Jake Ehlinger, younger brother of former Longhorns quarterback Sam Ehlinger, was found dead on May 6, Austin police said. Austin American-Statesman



SHIFT is ready for a different conversation.



For decades substance use and college campuses have been talked about as an inevitable rite of passage for college students, creating a norm that can far out shadow the dynamic pursuits of college students that revolve around academics and future opportunities.

SHIFT engages the community in dialogue that changes the culture of campus substance use from one of misuse to one of well-being.



NARCAN Nasal Spray 4 mg, Emergency Treatment of...



Images may be subject to copyright. Learn More

SHIFT Pilots Operation Naloxone The SHIFT List Events Innovation Funds Partners Get Involved Resources

Should I carry naloxone?

Sometimes it can be difficult to imagine how one person can have an impact on the culture of substance use – but you can! By carrying naloxone and learning the right way to administer it to someone having an opioid overdose, you have the potential to save a life. Even if you don't personally know anyone using opioids, you may find yourself in a situation where having naloxone on hand could make a huge difference. By showing that you care and taking the time to learn, you're helping to raise awareness about how important it is for each of us to play a part in shifting the culture of substance use.

How do I use naloxone?

Okay, so now you know what naloxone is, and why it's so important – but how do you actually use it on someone experiencing an opioid overdose? Great question – luckily, Operation Naloxone at UT provides free trainings for students, staff, and faculty. Email shift@utexas.edu for more information. Request an Operation Naloxone training.

How do you administer the nasal spray version of naloxone (Narcan)?

- Open the Narcan package, place the nozzle in the person's nostril and press the plunger.
- View the CDC video on how to administer Narcan.

Where can I find naloxone?

- Naloxone is available for distribution to all students, faculty, and staff at the Perry-Casteneda Library security front desk, the Longhorn Wellness Center (SSB 1.106), and the Center for Students in Recovery (Belmont 222).
- Naloxone is available for emergency access at all residence hall front desks, Sid Richardson Library, the Life Sciences Library, the Perry-Casteneda Library, and through UTPD.
- Many pharmacies dispense naloxone without a prescription, but there may be a copay depending on the insurer. You can call your insurance provider in advance to learn more about the potential copay cost.
- In Texas, you can request free naloxone via mail at MoreNarcanPlease.com.
- Map of free naloxone access sites in Austin.